Government of Karnataka Commissionerate of Pre-University Education II PUC Chemistry Practicals

EXPERIMENTS FOR CHEMISTRY PRACTICAL EXAMINATION

Time: 2 Hrs. Total Marks: 30

Q-I	Salt analysis	10				
	Analyse the given simple inorganic salt systematically	marks				
	and report one acid radical and one basic radical .					
Q-II	Titration (Volumetric Analysis)					
	Estimate the Molarity of KMnO ₄ solution using given					
	standard (0.1M) FAS solution.					
	(procedure of the titration should be given).	marks				
Q-III	Viva on tests for functional groups in organic					
	compounds:	4 marks				
IV	Submission of the duly completed and certified record	6 marks				
	TOTAL	30				
	TOTAL	marks				

SCHEME OF VALUATION

Time: 2 Hrs. Total Marks: 30

Q-I	Salt an	alysis (10 Marks)	Total Marks.		
	i) Preli	minary tests (any two correct)	1 mark		
	ii) Detection of Acid radical (4 Marks)				
	Group detection				
	(correct group identification – 1 mark				
	corr	ect radical identification – 1 ma	ark) 2 marks		
	Con	firmatory test	2 marks	10	
	iii) Detection of Basic radical (4 Marks)				
	Gro	ap detection			
	(correct group identification – 1 mark				
	correct radical identification – 1 mark) 2 mar				
	Con	firmatory test	2 marks		
	For writing systematic procedure with absence of previous				
	grou	ips	1 mark		
Q-II	Titrati	on (10 Marks)			
	i) For p	performing the experiment	3 marks		
	For recording the readings in the tabular column 1 mark				
	ii) For accuracy of the Titre value 3 marks				
	up to \pm 0.3 mL error 3 mark				
	± 0.4 mL error 2 marks				
	±	0.5 mL	1 mark		
	≥	0.6 mL & above	0 mark		
	iii) Calculations of Molarity (2 marks)				
	a. Formula 1 mark				
	b	. Substitution and answer (1+	1) 2 mark	marks	
Q-III	Viva or	compound (2 marks)			
	Four questions, two each on				
	any two functional groups (1x4)			marks	
IV	Record				
	Submis	ssion of the duly completed and	d certified record	marks	
	Sl.No	% of experiments performed	Maximum marks to		
		and recorded	be awarded		
	1	≥ 91%	6		
	2	≥81% to 90%	5		
	3	≥71% to 80%	4		
	4	Between 41% and 70%	3	00	
	5	40% & below 40%	0	30 marks	
	TOTAL				

Note:

- a) The following salts are suggested to be given for analysis for practical examination: NH₄Br, NH₄Cl, Al₂(SO₄)₃, MnSO₄, ZnSO₄/ ZnCO₃, CaCO₃, BaCl₂/ Ba(NO₃)₂, Sr(NO₃)₂/ SrCl₂, MgSO₄/ MgCO₃.
- **b)** For viva:

Functional group Tests

Alcohol Ceric ammonium nitrate test and Lucas test
Phenol Neutral ferric chloride, phthalein dye test
Aldehydes and ketones 2, 4 – DNPH and Tollen's reagent test

Carboxylic acid Litmus test, sodium bicarbonate test, esterification

Primary amine Carbylamine test, azo dye test

c) Inorganic salts and test for **organic compounds** other than the mentioned above but given in the prescribed manual can be given to students in regular practical class for practice.

d) All experiments as mentioned in the II PUC practical manual are to be conducted and recorded.